

Sol-gel prepared nanopowders of new mixed cobaltites-ferrites $\text{DyCo}_{1-x}\text{Fe}_x\text{O}_3$

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Nanocrystalline powders of mixed cobaltites-ferrites $\text{DyCo}_{1-x}\text{Fe}_x\text{O}_3$ ($x = 0.3, 0.5$ and 0.7) were prepared for the first time by sol-gel citrate method from Dy_2O_3 , $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ and $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ as initial reagents. Aqueous solutions of metal nitrates and citric acid (CA) taken in the molar ratio $n(\text{Dy}^{3+}) : n(\text{Co}^{2+}) : n(\text{Fe}^{3+}) : n(\text{CA}) = 1 : (1-x) : x : 4$ were mixed together on magnetic stirring for 30 min, gelled at 373–393 K for 4 h, after that head treated at 573 K for 1 h to obtain dry precursors. Finally as-obtained products were calcined at 1073 K for 2 h.

X-ray diffraction examinations revealed formation of pure perovskite structures isotypic with GdFeO_3 . No traces of parasitic phases were detected. Refined values of the lattice parameters of $\text{DyCo}_{1-x}\text{Fe}_x\text{O}_3$ prove the formation of continuous solid solution with orthorhombic perovskite structure (space group *Pbnm*) in DyCoO_3 – DyFeO_3 pseudobinary system. Average grain size of $\text{DyCo}_{1-x}\text{Fe}_x\text{O}_3$ powders, estimated from the analysis of XRD line broadening, was in the limit of 50–120 nm, depending on composition. Scanning electron microscopy of $\text{DyCo}_{0.7}\text{Fe}_{0.3}\text{O}_3$ sample revealed a lacy morphology of the powder consisting of irregular shaped 60–100 nm nanoparticles.

In comparison with a traditional energy- and time-consuming high-temperature solid-state synthesis of the mixed rare earth cobaltites-ferrites, required long-term sintering at 1473–1573 K with several intermediate regrindings, the low-temperature sol-gel citrate method is a very promising tool for the production of fine powders of $\text{RCo}_{1-x}\text{Fe}_x\text{O}_3$ perovskites, free of contamination of constituent metal oxides or other parasitic phases.

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